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Solution: Rearrange $Q = mc\Delta t$ to solve for T
 $Q = mc\Delta T$
 $8.7 \times 10^2 \text{ J} = (0.05 \text{ kg})(1.3 \times 10^2 \text{ J/kg}^\circ\text{C}) \Delta T$
 $8.7 \times 10^2 \text{ J} = 6.5 \text{ J}^\circ\text{C} \Delta T$

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$$iC) 8.7 \times 10^2 \text{ J } 6.5 \text{ J } \frac{1}{4}C = T \text{ 2) } 22 \text{ iC } T \text{ 2}$$
$$= 134 \text{ iC} + 22 \text{ iC} = 156 \text{ iC } T \text{ 2} = 160 \text{ iC}$$

Statement: The final temperature of the lead was 160°C .

Chapter 6 Review, Understanding pages 310-315

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